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UTILITY PATENT APPLICATION TRANSMITTAL

(Only for new nonprovisional applications under 37 CFR 1.53(b))

Attorney Docket No.	Beiersdorf 670
First Inventor	Gerhard BENNER
Title	See Appendix
Express Mail Label No.	EL 770070055 US

APPLICATION ELEMENTS

See MPEP chapter 600 concerning utility patent application contents.

ADDRESS TO:

Assistant Commissioner for Patents
Box Patent Application
Washington, DC 20231

1. ☒ Fee Transmittal Form (e.g. PTO/SB/17)
(Submit an original and a duplicate for fee processing)
2. ☐ Applicant claims small entity status.
See 37 CFR 1.27.
3. ☒ Specification [Total Pages 41]
(Preferred arrangement set forth below)
 - Descriptive title of the invention
 - Cross Reference to Related Applications
 - Statement Regarding Fed sponsored R & D
 - Reference to sequence listing, a table, or a computer program listing appendix
 - Background of the Invention
 - Brief Summary of the Invention
 - Detailed Description of the Drawings (if filed)
 - Claim(s)
 - Abstract of the Disclosure
4. ☒ Drawing(s) (35 U.S.C. 113) [Total Sheets 4]
5. ☒ Oath or Declaration [Total Pages 2]
 - a. ☒ Newly executed (original or copy)
 - b. ☐ Copy from a prior application (37 CFR 1.63 (d))
(For continuation/divisional with Box 17 completed)
 - i. ☐ DELETION OF INVENTOR(S)
Signed Statement attached deleting inventor(s)
Named in the prior application, see 37 CFR 1.63 (d)(2) and 1.33(b).
6. ☐ Application Data Sheet. See 37 CFR 1.76

7. ☐ CD-ROM or CD-R in duplicate, large table or Computer Program (Appendix)
8. ☐ Nucleotide and/or Amino Acid Sequence Submission
(If applicable, all necessary)
 - a. ☐ Computer Readable Forms (CRF)
 - b. Specification Sequence Listing on:
 - i. ☐ CD-ROM or CD-R (2 copies); or
 - ii. ☐ paper
 - c. ☐ Statements verifying identity of above copies

ACCOMPANYING APPLICATION PARTS

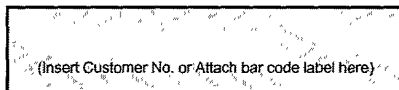
9. ☒ Assignment Paper (cover sheet & document(s))
10. ☐ 37 CFR 3.73(b) Statement ☒ Power of Attorney
(When there is an assignee)
11. ☐ English Translation Document (if applicable)
12. ☒ Information Disclosure ☒ Copies of IDS Citations (10)
Statement (IDS)/PTO-1449
13. ☒ Preliminary Amendment
14. ☒ Return Receipt Postcard (MPEP 503)
(Should be specifically itemized)
15. ☒ Certified Copy of Priority Document(s)
(If foreign priority is claimed)
16. ☒ Other: Appendix, German Search Report of June 21, 2000

17. If a CONTINUING APPLICATION, check appropriate box, and supply the requisite information below and in preliminary amendment, or in an Application Data Sheet under 37 CFR 1.76:

☐ Continuation ☐ Divisional ☐ Continuation-in-part (CIP) of prior application No: _____
Prior application information: Examiner _____ Group / Art Unit: _____

For CONTINUATION OR DIVISIONAL APPS only: The entire disclosure of the prior application, from which an oath or declaration is supplied under Box 5b, is considered a part of the disclosure of the accompanying continuation or divisional application and is hereby incorporated by reference. The incorporation can only be relied upon when a portion has been inadvertently omitted from the submitted application parts.

18. CORRESPONDENCE ADDRESS

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**FEE TRANSMITTAL
for FY 2001**

Patent fees are subject to annual revision.

TOTAL AMOUNT OF PAYMENT

(\$) 710.

Complete if Known

Application Number	To Be Assigned
Filing Date	Herewith
First Named Inventor	Gerhard BENNER et al.
Examiner Name	To Be Assigned
Group Art Unit	To Be Assigned
Attorney Docket No.	Beiersdorf 670

METHOD OF PAYMENT

1. ☒ The Commissioner is hereby authorized to charge indicated fees and credit any overpayments to:

Deposit Account Number: **NORRIS, McLAUGHLIN & MARCUS, P.A.**

Deposit Account Name: **14-1263**

☒ Charge Any Additional Fee Required Under 37 CFR 1.16 and 1.17

☐ Applicant claims small entity status. See 37 CFR 1.27

2. ☐ Payment Enclosed:

☐ Check ☐ Credit card ☐ Money Order ☐ Other

FEE CALCULATION

1. BASIC FILING FEE

Large Entity Fee Code (\$)	Small Entity Fee Code (\$)	Fee Description	Fee Paid
101 710	201 355	Utility filing fee	710.
106 320	206 160	Design filing fee	
107 490	207 245	Plant filing fee	
108 710	208 355	Reissue filing fee	
114 150	214 75	Provisional filing fee	

SUBTOTAL (1) (\$) 710.

2. EXTRA CLAIM FEES

Total Claims	Extra Claims	Fee from below	Fee Paid
9	-20** = 0	X	0
1	-3** = 0	X	0
Multiple Dependent			0

Large Entity Fee Code (\$)	Small Entity Fee Code (\$)	Fee Description
103 18	203 9	Claims in excess of 20
102 80	202 40	Independent claims in excess of 3
104 270	204 135	Multiple dependent claim, if not paid
109 80	209 40	** Reissue independent claims over original patent
110 18	210 9	** Reissue claims in excess of 20 and over original patent

SUBTOTAL (2) (\$) 0.

**or number previously paid, if greater; For Reissues, see above

FEE CALCULATION (continued)

3. ADDITIONAL FEES

Large Entity Fee Code (\$)	Small Entity Fee Code (\$)	Fee Description	Fee Paid
105 130	205 65	Surcharge - late filing fee or oath	
127 50	227 25	Surcharge - late provisional filing fee or cover sheet	
139 130	139 130	Non-English specification	
147 2,520	147 2,520	For filing a request for ex parte reexamination	
112 920*	112 920*	Requesting publication of SIR prior to Examiner action	
113 1,840*	113 1,840*	Requesting publication of SIR after Examiner action	
115 110	215 55	Extension for reply within first month	
116 390	216 195	Extension for reply within second month	
117 890	217 445	Extension for reply within third month	
118 1,390	218 695	Extension for reply within fourth month	
128 1,890	228 945	Extension for reply within fifth month	
119 310	219 155	Notice of Appeal	
120 310	220 155	Filing a brief in support of an appeal	
121 270	221 135	Request for oral hearing	
138 1,510	138 1,510	Petition to institute a public use proceeding	
140 110	240 55	Petition to revive - unavoidable	
141 1,240	241 620	Petition to revive - unintentional	
142 1,240	242 620	Utility issue fee (or reissue)	
143 440	243 220	Design issue fee	
144 600	244 300	Plant issue fee	
122 130	122 130	Petitions to the Commissioner	
123 50	123 50	Petitions related to provisional applications	
126 240	126 240	Submission of Information Disclosure Stmt	
581 40	581 40	Recording each patent assignment per property (times number of properties)	
146 710	246 355	Filing a submission after final rejection (37 CFR § 1.129(a))	
149 710	249 355	For each additional invention to be examined (37 CFR § 1.129(b))	
179 710	279 355	Request for Continued Examination (RCE)	
169 900	169 900	Request for expedited examination of a design application	

Other fee (specify) _____

* Reduced by Basic Filing Fee Paid

SUBTOTAL (3) (\$)

SUBMITTED BY

Complete (if applicable)

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DATE OF DEPOSIT: November 27, 2000
EXPRESS MAIL LABEL NO.: EL 770070055 US
Our Docket Number Beiersdorf 670-VMM
Client Docket Number 6713-Dr. Wi-ar

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

APPLICANTS : Dr. Gerhard Benner et al.
SERIAL NUMBER : To be assigned
FILED : Herewith
FOR : COSMETIC AND DERMATOLOGICAL LIGHT PROTECTION
FORMULATIONS IN THE FORM OF O/W
MACROEMULSIONS OR O/W MICROEMULSIONS
CONTAINING SHEA BUTTER
ART UNIT : To be assigned
EXAMINER : To be assigned

November 27, 2000

PRELIMINARY AMENDMENT

Hon. Commissioner of Patents
Washington, DC 20231

Sir:

Applicants submit the following amendments for entry and consideration in order to place the instant application in better condition for examination on the merits.

IN THE CLAIMS:

Please amend the following claims:

00/231/5872/60

In Claim 1, line 1, delete “, in particular O/W microemulsion” and
insert therein -- comprising-- .

In Claim 1, line 2 delete “comprising”.

In Claim 1, line 14, delete “W/O emulsifiers,” and insert therein -- oil-in-water
emulsifiers, and-- .

In Claim 2, line 1, delete “O/W macroemulsion or O/W microemulsionen” and insert
therein --oil-in-water emulsion-- .

In Claim 2, line 2, delete “or the emulsifiers A is or are present in concentrations” and
insert therein – is present in a concentration-- .

In Claim 2, line 3, delete “, preferably 0.05-10% by weight, particularly preferably 0.1-
5% by weight, in each case”

In Claim 3, line 1, delete “O/W macroemulsion or O/W microemulsion” and insert
therein --oil-in-water emulsion-- .

In Claim 3, line 2, delete “, advantageously 0.5 to 10% by weight, very particularly
preferably 1 to 5% by weight,”

Please add the following new claims:

-- Claim 4 (New) The oil-in-water emulsion of claim 1, wherein the oil-in-water emulsion is a macroemulsion. --

-- Claim 5 (New) The oil-in-water emulsion of claim 1, wherein the oil-in-water emulsion is a microemulsion. --

-- Claim 6 (New) The oil-in-water emulsion of claim 2, wherein the emulsifier A is present in a concentration of 0.05-10% by weight based on the total weight of the composition. --

-- Claim 7 (New) The oil-in-water emulsion of claim 2, wherein the emulsifier A is present in a concentration of 0.1-5% by weight based on the total weight of the composition.--

-- Claim 8 (New) The oil-in-water emulsion of claim 3, which comprises 0.5 to 10% by weight of shea butter. --

-- Claim 9 (New) The oil-in-water emulsion of claim 3, which comprises 1 to 5% by weight of shea butter. --

REMARKS

Claims 1 - 3 are amended and new Claims 4 - 9 added to more clearly delineate the subject matter to be claimed and to correct minor typographical/ grammatical errors. Applicants aver that no forbidden new matter has been added to the application. Claims 1-9 are the only claims pending in the instant application and are hereby presented for consideration.

Respectfully submitted,

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Description

10 Cosmetic and dermatological light protection formulations in the form of
O/W macroemulsions or O/W microemulsions containing shea butter

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The present invention relates to cosmetic and dermatological emulsions, in particular
skincare cosmetic and dermatological emulsions. In an advantageous embodiment, the
present invention relates to a use which permits the preparation of cosmetically elegant
15 preparations, in particular emulsions, having excellent sensory and skincare properties.

The skin is the largest human organ. Amongst its many functions (for example for
temperature regulation and as a sensory organ) the barrier function, which prevents the
skin (and ultimately the entire organism) from drying out, is doubtless the most important.
20 At the same time, the skin acts as a protective device against the penetration and
absorption of external substances. This barrier function is effected by the epidermis,
which, as the outermost layer, forms the actual protective sheath against the
environment. Being about one tenth of the total thickness, it is also the thinnest layer of
the skin.

25 The epidermis is a stratified tissue in which the outer layer, the horny layer (Stratum
corneum), is the part which is of significance for the barrier function. The Elias skin
model, which is currently recognized in the specialist-field (*P. M. Elias, Structure and
Function of the Stratum Corneum Permeability Barrier, Drug Dev. Res. 13, 1988, 97-*
30 *105*), describes the horny layer as a two-component system, similar to a brick wall (bricks
and mortar model). In this model, the horny cells (corneocytes) correspond to the bricks,
and the lipid membrane in the intercellular spaces, which is of complex composition,
corresponds to the mortar. This system is essentially a physical barrier to hydrophilic

substances, but, because of its narrow and multilayered structure, can equally, however, also only be passed by lipophilic substances with difficulty.

5 The present invention relates, in a particular embodiment, to cosmetic or pharmaceutical preparations having a reduced feeling of stickiness, to processes for their preparation, and also to the use of active ingredients for reducing the feeling of stickiness of cosmetic preparations.

10 Apart from its barrier action against external chemical and physical influences, the epidermal lipids also contribute to the holding together of the horny layer and have an effect on the smoothness of the skin. In contrast to the sebaceous gland lipids, which do not form a continuous film on the skin, the epidermal lipids are distributed over the entire horny layer.

15 The extremely complex interaction of the moisture-binding substances and of the lipids of the upper layers of the skin is very important for the regulation of skin moisture. For this reason, cosmetics generally comprise, in addition to balanced lipid mixtures and water, water-binding substances.

20 In addition to the chemical composition, however, the physical behavior of these substances is also of importance. The development of very biocompatible emulsifiers and surfactants is therefore desirable. Products formulated therewith aid the liquid-crystalline organization of the intercellular lipids of the Stratum corneum, thereby improving the barrier properties of the horny layer. It is particularly advantageous if their molecular
25 constituents consist of substances which are naturally occurring in the epidermis.

Cosmetic skin care primarily means that the natural function of the skin as a barrier against environmental influences (e.g. dirt, chemicals, microorganisms) and against the loss of endogenous substances (e.g. water, natural fats, electrolytes) is strengthened or
30 rebuilt.

If this function is impaired, increased resorption of toxic or allergenic substances or attack by microorganisms may result, leading to toxic or allergic skin reactions.

Another aim of skin care is to compensate for the loss by the skin of lipids and water caused by daily washing. This is particularly important when the natural regeneration ability is insufficient. Furthermore, skincare products should protect against environmental influences, in particular against sun and wind, and delay skin aging.

- 5 Medicinal topical compositions generally comprise one or more medicaments in an effective concentration. For the sake of simplicity, in order to distinguish clearly between cosmetic and medicinal use and corresponding products, reference is made to the legal provisions of the Federal Republic of Germany (e.g. Cosmetics Directive, Foods and Drugs Act).

10

Customary cosmetic forms of application are emulsions. This term generally means a heterogeneous system of two liquids which are immiscible or miscible only to a limited extent with one another, which are usually referred to as phases. One is in the form of droplets (disperse or internal phase), while the other liquid forms a continuous (coherent or internal) phase. Less common forms of application are multiple emulsions, i.e. those which, in the droplets of the dispersed (or discontinuous) phase, comprise for their part droplets of a further dispersed phase, e.g. W/O/W emulsions and O/W/O emulsions.

15

20

More recent findings have recently led to a better understanding of cosmetic emulsions which are of relevance in practice. Here, it is assumed that the emulsifier mixtures used in excess form lamellar liquid-crystalline phases or crystalline gel phases. In the gel network theory, stability and physicochemical properties of such emulsions are attributed to the formation of viscoelastic gel networks.

25

If the two liquids are water and oil and the oil droplets are finely dispersed in water, then this is an oil-in-water emulsion (O/W emulsion, e.g. milk). The basic character of an O/W emulsion is defined by the water. In the case of a water-in-oil emulsion (W/O emulsion, e.g. butter) the principle is reversed, the basic character being determined here by the oil.

30

In order to be able to ensure the metastability of emulsions, interface-active substances, i.e. emulsifiers, are usually necessary. The use per se of customary cosmetic emulsifiers is entirely acceptable. Nevertheless, emulsifiers, as ultimately any chemical substance,

may in certain cases cause allergic reactions or reactions based on oversensitivity of the user. For example, it is known that in some particularly sensitive people, certain light dermatoses are triggered by certain emulsifiers and simultaneous action of sunlight.

- 5 It is possible to prepare emulsifier-free preparations which, for example, have, in an aqueous phase, dispersed oil droplets, similar to an O/W emulsion. A prerequisite for this may be that the continuous aqueous phase has a gel framework which stabilizes the dispersed phase, and other conditions besides. Such systems are sometimes called hydrodispersions or oleodispersions depending on which is the disperse phase and
10 which is the continuous phase.

- For cosmetics technology, however, it is neither necessary nor possible to dispense with emulsifiers altogether, especially since there is a certain choice of particularly mild emulsifiers. However, the prior art lacks a satisfactorily broad range of such emulsifiers
15 which would then also significantly broaden the application spectrum of correspondingly mild cosmetic preparations which are tolerated by the skin.

- An object of the present invention was therefore to provide cosmetic and dermatological preparations having excellent skincare properties.
20

- A disadvantage, in particular of O/W emulsions, is often their inadequate stability to relatively high electrolyte concentrations, which manifests itself in phase separation. This can indeed sometimes lead to problems, even in the case of W/O emulsions, although this is by no means as important here as in the case of O/W systems. Although these
25 can often be remedied to a certain extent through appropriate choice of the emulsifier system, other disadvantages, however, arise just as often.

- On the other hand, it is often desirable to use certain electrolytes in order to be able to utilize their other physical, chemical or physiological properties.
30

The concentrations of all of the constituents of a cosmetic or dermatological preparation are usually given in units such as % by weight, mol% and the like. In view of their greater or lesser dissociation into cations and anions, often in several dissociation stages, it sometimes appears more advantageous, for the description of the present invention and

its technical background, to start from the ionic strength of a given electrolyte in its solution.

The ionic strength I of an electrolyte solution is defined as

$$I = \frac{1}{2} \sum_i c_i z_i^2$$

where c_i are the concentrations of the individual types of ion (in mol/l) and z_i are their charges. The physical unit of ionic strength is that of a concentration (mol/l).

For example, a 1% strength (= 0.17 molar) sodium chloride solution has an ionic strength $I = 0.17$.

Another object of the present invention was therefore to discover ways of producing cosmetic or dermatological emulsions, in particular O/W emulsions, which are stable to increased electrolyte concentrations – or increased ionic strengths.

The person skilled in the art is naturally aware of a large number of ways of formulating stable O/W preparations for cosmetic or dermatological use, for example in the form of creams and ointments, which are spreadable in the range from room temperature to skin temperature, or as lotions and milks, which are more likely to be flowable in this temperature range. However, there are only a few formulations in the prior art which are of sufficiently low viscosity that they would, for example, be sprayable.

In addition, low-viscosity preparations of the prior art frequently have the disadvantage that they are unstable, and are limited to a narrow field of application or to a limited choice of feed material. Low-viscosity products in which, for example, highly polar oils – such as the vegetable oils otherwise frequently used in commercially available products – are sufficiently stabilized are therefore currently not on the market.

The term “viscosity” means the property of a liquid to resist the mutual laminar displacement of two neighboring layers (internal friction). This so-called dynamic viscosity is nowadays defined according to $\eta = \tau / D$ as the ratio of shear stress to the velocity gradient perpendicular to the direction of flow. For Newtonian liquids, η is a material constant having the SI unit Pascal second (Pa·s) at a given temperature.

The quotient $\nu = \eta / \rho$ from the dynamic viscosity η and the density ρ of the liquid is referred to as the kinematic viscosity ν and is given in the SI unit m^2/s .

- 5 Fluidity (ϕ) is the inverse of viscosity ($\phi = 1/\eta$). In the case of ointments and the like, the use value is codetermined by the so-called tack. The tack of an ointment or ointment base or the like means its property to draw threads of varying lengths when a small sample is removed; accordingly, a distinction is made between short- and long-stretch substances.

10

While the graphical representation of the flow behavior of Newtonian liquids at a given temperature produces a straight line, in the case of so-called non-Newtonian liquids considerable deviations often arise, depending on the particular velocity gradient D (shear rate $\dot{\gamma}$) or the shear stress τ . In these cases, the so-called apparent viscosity can

- 15 be determined which, although not bound to the Newtonian equation, can be used to determine the true viscosity values by graphical methods.

Falling body viscometry is suitable only for investigating Newtonian liquids and gases. It is based on Stokes' law, according to which, for the falling of a sphere through a liquid

- 20 which flows around it, the dynamic viscosity η can be determined from

$$\eta = \frac{2r^2(\rho_K - \rho_{Fl}) \cdot g}{9 \cdot v}$$

where

r = radius of the sphere, v = fall velocity, ρ_K = density of the sphere, ρ_{Fl} = density of the liquid and g = acceleration of the fall.

25

O/W emulsions with a low viscosity which have a storage stability as is required for marketable products can only be formulated in accordance with the prior art in a very complex manner. Accordingly, the supply of such formulations is extremely low. Nevertheless, formulations of this type could offer the consumer hitherto unknown

30

cosmetic results.

An object of the present invention was to make available preparations which have very low viscosity and do not have the disadvantages of the prior art.

For polyol fatty acid esters, the definition of the HLB value is given by the formula I

$$\text{HLB} = 20 * (1 - S/A)$$

For a group of emulsifiers whose hydrophilic moiety consists only of ethylene oxide units,
5 the formula II applies

$$\text{HLB} = E/5$$

where S = saponification number of the ester,
 A = acid number of the recovered acid,
 E = mass fraction of ethylene oxide (in %) based on the overall
10 molecule.

Emulsifiers with HLB values of 6-8 are generally W/O emulsifiers, and those with HLB values of 8-18 are generally O/W emulsifiers.

15 Reference: "Kosmetik - Entwicklung, Herstellung und Anwendung kosmetischer Mittel" [Cosmetics - Development, Preparation and Use of Cosmetic Compositions], W. Umbach (Ed.), Georg Thieme Verlag 1988.

Hydrophilic emulsifiers (with high HLB values) are generally O/W emulsifiers.
20 Accordingly, hydrophobic or lipophilic emulsifiers (with low HLB values) are generally W/O emulsifiers.

US patent specification 4,931,210 describes a process for the preparation of W/O/W emulsions where polyglycerol polyricinoleates are used as emulsifiers.

25 The droplet diameters of customary "simple", i.e. non-multiple emulsions are in the range from about 1 μm to about 50 μm . Such "macroemulsions" are, without further coloring additives, milky-white in color and opaque. Finer "macroemulsions", the droplet diameters of which are in the range from about 10^{-1} μm to about 1 μm are, again without coloring
30 additives, bluish-white in color and opaque. Such "macroemulsions" usually have high viscosity.

It is only micellar and molecular solutions having particle diameters of less than about $10^2 \mu\text{m}$, but which are no longer to be regarded as true emulsions, which have a clear and transparent appearance.

- 5 By contrast, the droplet diameter of microemulsions is in the range from about $10^{-2} \mu\text{m}$ to about $10^{-1} \mu\text{m}$. Microemulsions are translucent and in most cases of low viscosity. The viscosity of many microemulsions of the O/W type is comparable with that of water.

10 The advantage of microemulsions is that, in the disperse phase, active ingredients can be present in essentially more finely dispersed form than in the disperse phase of "macroemulsions". A further advantage is that they are sprayable as a result of their low viscosity. If microemulsions are used as cosmetics, corresponding products are characterized by high cosmetic elegance.

- 15 It is known that hydrophilic emulsifiers change their solubility behavior from water-soluble to fat-soluble with increasing temperature. The temperature range in which the emulsifiers have changed their solubility is called the phase inversion temperature range (PIT).

- 20 T.J.Lin, H.Kurihara and H.Ohta (Journal of the Society of Cosmetic Chemists 26, pp. 121 - 139, March 1975) show that for nonpolar oils extremely unstable multiple emulsions may be present in the PIT range.

The object of the present invention was therefore to remedy these shortcomings.

25

It was furthermore an object of the present invention to make available preparations which significantly improve the condition of the skin, in particular reduce skin roughness.

- 30 It is admittedly known that certain substances, for example a few selected powder raw materials, in particular talc, can be added to reduce a feeling of stickiness and also a feeling of greasiness. However, apart from the fact that this is only rarely completely successful, such an addition also changes the viscosity of the product in question and lowers the stability.

The object was therefore to remedy all of these disadvantages of the prior art. In particular, the aim was to make available products with reduced stickiness or greasiness. Products in the field of care cosmetics, decorative cosmetics and pharmacological technology were likewise to be freed from the described disadvantages of the prior art.

5

Furthermore, it was an object of the invention to develop cosmetic bases for cosmetic preparations which are characterized by good skin compatibility.

10 It was also an object of the present invention to provide product with the widest possible variety of applications. For example, the aim was to provide bases for preparation forms such as cleansing emulsions, face and bodycare preparations, but also, in particular, medicopharmaceutical application forms, for example preparations against acne and other skin conditions.

15 Surprisingly, we have found, and herein lies the basis to the attainment of the objects, that oil-in-water emulsions, in particular O/W microemulsions

(a) comprising at least one emulsifier (emulsifier A), chosen from the group of emulsifiers having the following properties

20 - their lipophilicity is either dependent on the pH inasmuch as an increase or decrease in pH results in an increase or decrease in lipophilicity, it being unimportant which of the two possibilities of change in the lipophilicity is effected by the increase or the decrease in the pH, and/or

25 - their lipophilicity is dependent on the temperature inasmuch as the lipophilicity increases with increasing temperature and their hydrophilicity increases with decreasing temperature,

(b) also optionally further substances which are soluble or dispersible in the oil phase or the water phase, including, preferably, those chosen from the group of emulsifiers not covered by the definition of emulsifier A, in particular those which act primarily as W/O emulsifiers,

30 (c) an effective amount of shea butter,
overcome the disadvantages of the prior art.

Shea butter (also: shea fat, karite fat, galam butter) is a natural solid obtained from the plant *Butyrospermum parkii*, the African shea butter tree and is available in commercial

quantities. Usually, shea fat comprises 89 to 98% by weight of triglycerides, glycerol partial esters and free fatty acids, and a content of from 2 to 11% by weight of nonhydrolyzable fractions, of which hydrocarbons ("karitenes"), triterpene alcohols and sterols are the most important.

5

If the nonhydrolyzable fractions are alcohols, the majority are present in the form of cinnamic esters. Accordingly, the designation "nonhydrolyzable" for this fraction does not apply in the strict sense. From the overview literature on the theme of shea fat, reference may be made to the publications by Itoh et al. in *Oleagineux*, 29,5, 253 (1974), Olberg et al. in *Seifen-Öle-Fette-Wachse*, 113, 10, 333 (1987) and Peers in *J.Sci.Food Agric.* 28, 1000 (1977).

10

Shea fat and shea fat fractions have been known in skincare for a long time for their care and protecting properties. The action of these fatty substances is attributed to an influencing of biochemical inflammation processes since the UV absorption in the UVA and UVB region is only weakly defined. Nevertheless, the prior art was unable to point the way to the present invention.

15

If phase inversion, within the meaning of the present invention, is essentially triggered by varying the temperature, O/W emulsions, in particular O/W microemulsions, are obtainable, where the size of the oil droplets is essentially determined by the concentration of the emulsifier(s) used, inasmuch as a higher emulsifier concentration brings about smaller droplets and a lower emulsifier concentration leads to relatively large droplets. If phase inversion is essentially triggered by varying the temperature, it is entirely advantageous to dispense with further emulsifiers not covered by the definition of emulsifier A, namely W/O emulsifiers.

20

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According to the invention, cosmetic or dermatological preparations advantageously comprise 0.1 to 20% by weight, advantageously 0.5 to 10% by weight, very particularly preferably 1 to 5% by weight, of shea butter.

30

If phase inversion is essentially triggered by varying the pH, O/W emulsions, in particular O/W microemulsions are obtainable. If phase inversion is triggered essentially by varying

the pH, it is entirely advantageous to use one or more further emulsifiers not covered by the definition of emulsifier A, namely W/O emulsifiers.

5 According to the invention, O/W microemulsions can be obtained if the oil phase fraction is less than about 20% by weight, in particular less than about 15% by weight, based on the overall weight of the preparation, if less than about 5% by weight of an additional W/O emulsifier not covered by the definition of the emulsifier A is present, and/or if the oil phase has a high proportion of polar oils.

10 According to the invention, O/W emulsions ("macroemulsions") can be obtained if less than about 5% by weight of an additional W/O emulsifier not covered by the definition of emulsifier A and more than about 20% by weight of a polar oil phase are present. Additional gel formers (e.g. carbopols, xanthan gum, cellulose derivatives) can advantageously be used.

15 In individual cases it is possible to slightly exceed or fall below the abovementioned concentration limits and nevertheless obtain the emulsion types in question. In view of the wide diversity of suitable emulsifiers and oil constituents, this is not unexpected for the person skilled in the art, so that he or she knows that such excesses or deficits do not
20 depart from the basis of the present invention.

Where the phase inversion is initiated essentially by varying the temperature, O/W emulsions, especially O/W microemulsions, are obtainable, the size of the oil droplets being determined essentially by the concentration of the emulsifier or emulsifiers used,
25 such that a higher emulsifier concentration produces smaller droplets and a lower emulsifier concentration leads to larger droplets. If phase inversion is triggered essentially by varying the temperature, it is entirely advantageous, although not absolutely necessary, to dispense with further emulsifiers not covered by the definition of emulsifier A, namely W/O emulsifiers.

30 If phase inversion is essentially triggered by varying the pH, O/W emulsions, in particular O/W microemulsions, and also O/W/O emulsions, are obtainable. If phase inversion is triggered essentially by varying the pH, it is entirely advantageous to use one or more further emulsifiers not covered by the definition of emulsifier A, namely W/O emulsifiers.

According to the invention, O/W microemulsions can be obtained if the oil phase fraction is less than about 20% by weight, in particular less than about 15% by weight, based on the overall weight of the preparation, if less than about 5% by weight of an additional
5 W/O emulsifier which is not covered by the definition of emulsifier A is present, and/or if the oil phase has a high proportion of polar oils.

According to the invention, O/W emulsions ("macroemulsions") can be obtained if less than about 5% by weight of an additional W/O emulsifier not covered by the definition of
10 emulsifier A and more than about 20% by weight of a polar oil phase are present. Additional gel formers (e.g. carbopols, xanthan gum, cellulose derivatives) can advantageously be used.

In individual cases it is possible to slightly exceed or fall below the abovementioned
15 concentration limits and nevertheless obtain the emulsion types in question. In view of the wide diversity of suitable emulsifiers and oil constituents, this is not unexpected for the person skilled in the art, so that he or she knows that such excesses or deficits do not depart from the basis of the present invention.

Surprisingly, we have found that the pigment particle(s) used according to the invention are in the form of solids, and to a certain extent "encapsulated", namely separate from other constituents of the preparations, in some of which they can even have limited solubility. It is assumed that the solid particles of the sparingly soluble UV filter substances receive a coating film as a result of the incorporation process according to
20 the invention, which film presumably comprises emulsifier molecules as essential constituent.
25

According to the invention the recrystallization of the s-triazine derivative(s) used according to the invention can be prevented. Moreover, light protection preparations are
30 obtainable according to the invention which have excellent use properties.

Fig. 1 shows a very simplified representation of a phase diagram. The variable parameter P is plotted against the temperature θ as second variable. P is here a concentration parameter, either the proportion of the oil phase, the proportion of the water phase or the

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concentration of an emulsifier or an emulsifier mixture. For systems according to the invention it is the case that at relatively low temperatures an O/W emulsion is present and as the temperature increases the phase inversion range can be passed through. If the temperature is increased further, W/O emulsions are observed. The structure of the system in the phase inversion range is seemingly unimportant for the present invention. For example, it is conceivable that lamellar phases, bicontinuous phases, cubic, hexagonal or inverse hexagonal phases are present in the phase inversion range, and also that the phase inversion range is composed of two or more identical or more or less different phases.

The phase inversion range can be represented mathematically as a point quantity within the straight-line coordinate system Σ , which is formed by the parameters of temperature, the concentration of a suitable emulsifier or of an emulsifier mixture in the preparation and the respective concentrations of the oil phase and water phase, according to:

$$\Sigma = \{O, \theta, m, H, W\},$$

where O = coordinate origin

θ = temperature

m = concentration of the emulsifier/emulsifier mixture

H = concentration of the oil phase

W = concentration of the water phase.

Strictly speaking of course, in a multicomponent emulsifier system, the contribution m_i of each individual emulsifier to the overall function must be taken into consideration which, in the case of an i-component emulsifier system, leads to the relationship

$$\Sigma = \{O, \theta, m_1, m_2, \dots, m_i, H, W\}.$$

The phase inversion range Φ here in the mathematical sense is a continuous region or a large number of continuous regions within the coordinate system Σ . Φ represents the total amount of coordinate points $K(\theta, a, m_1, m_2, \dots, m_i, H, W)$, which determine mixtures according to the invention of a water phase of concentration W, oil phase of concentration H, i emulsifiers according to the invention of concentration m_i at the

temperature θ , and for which, upon passing from a coordinate $K_1 \in \Phi$ to a coordinate $K_2 \in \Phi$, phase inversion occurs, as described in Fig. 2.

It is irrelevant here whether the phase inversion range of a given system is a single coherent $(i + 3)$ -dimensional field or consists of two or more such fields which are coherent but separate from one another, i.e. corresponding to two or more phase inversion ranges of a given system. Within the scope of the disclosure presented herein, "the" or "a" phase inversion range is always generally referred to, even if two or more such ranges separate from one another are present.

10

The variable coordinates given in Fig. 2 are temperature θ and the above-described concentration parameter P , it being possible for which specific concentration parameter is involved to remain open. On passing from K_1 to K_2 , only the temperature is increased, and the other variables are kept constant.

15

Under the conditions according to the invention, this process is not reversible, i.e. if the system reverts from the coordinate $K_2 \in \Phi$ to the coordinate $K_1 \in \Phi$, transparent O/W microemulsions according to the invention may be obtained.

20

The practice of preparing a microemulsion according to the invention accordingly advantageously consists, after choosing suitable raw materials, i.e. water phase and oil phase, one or more O/W emulsifiers used according to the invention, the latter being present in concentrations at which phase inversion is possible for the given mixture, and optionally further substances, in combining the individual components with stirring, bringing about a phase inversion by increasing the temperature of the mixture, and thereafter allowing the mixture to cool to room temperature with continued stirring.

25

However, it is also possible here to vary two or more parameters at the same time, as shown in Fig.3. In Fig. 3 the concentration of the water phase is plotted against the temperature. Starting from the coordinate $K_1 \in \Phi$, by increasing the temperature, while maintaining all other parameters, the coordinates $K_2 \in \Phi$ and $K_4 \in \Phi$ can be reached, or $K_3 \in \Phi$. Starting from the coordinates K_3 and K_4 , by lowering the temperature, while maintaining all other parameters, back to the coordinate K_1 , O/W microemulsions according to the invention can be obtained.

30

Starting from the coordinates K_3 and K_4 , by lowering the temperature, and by additionally varying the concentration of the oil phase, in Fig.3 by the addition of water, the coordinate K_5 can be reached and O/W microemulsions according to the invention can be obtained.

In view of Fig. 3, it is logical that starting from the coordinate K_4 , although this is outside the phase inversion range, systems similar to those which start from K_3 can be obtained, since starting from K_4 if the temperature is lowered, the phase inversion range must also automatically be traversed.

Also, starting from the coordinate K_1 , by varying the concentration of the water phase, i.e. for example by adding water, as is shown in Fig. 3, the coordinate K_5 can be reached, and O/W microemulsions according to the invention can be obtained. In this regard, however, it must first be mentioned that in this case an O/W microemulsion according to the invention, to a certain extent as a concentrate, must already be present, which is then converted into an O/W microemulsion according to the invention of different composition by dilution.

However, having said all that, it was surprising and therefore involves independent inventive activity, that starting from the coordinate K_2 , which lies outside the phase inversion range, either by simply varying the temperature back to the coordinate K_1 or by additionally varying the concentration of the oil phase, i.e., for example, by additional dilution with a water phase to the coordinate K_5 , O/W microemulsions according to the invention are also obtainable without passing through phase inversion. This is advantageously effected by bringing a mixture of the base components, comprising water phase, oil phase, one or more of the O/W emulsifiers used according to the invention, if desired one or more W/O emulsifiers, and optionally further auxiliaries, additives and/or active ingredients, which form an O/W emulsion below the phase inversion temperature range, to a temperature

- at which the components which are soluble in the oil phase are present either in dissolved form or at least in the molten state
- and which corresponds at least to the melting temperature of the highest-melting oily component which is not present in the dissolved state,

- which is below the phase inversion temperature range of the system, and afterwards cooling the resulting O/W emulsion to room temperature to form an O/W microemulsion. This is preferably carried out with stirring.

- 5 This process according to the invention is particularly suitable if heat-sensitive or readily volatile substances are to be incorporated into the O/W microemulsions according to the invention. Moreover, this process, which is carried out at relatively low temperatures, is energy-saving compared with customary processes.
- 10 Fig. 4 describes the case in which no O/W emulsifier according to the invention is initially present in the coordinate L_1 , and in which the system is brought to a coordinate $L_3 \notin \Phi$ or to a coordinate $L_2 \notin \Phi$ by increasing the temperature. The coordinate L_2 can of course also be achieved by cooling a system present in the coordinate L_3 . The coordinates L_2 and L_3 , in which, for example, W/O emulsions can be present, differ in principle merely by virtue of the fact that the temperature assigned to L_3 is higher than that temperature which can be assigned to the phase inversion temperature range.
- 15

- The presence of an additional W/O emulsifier for systems which are symbolized in Fig. 4 is not necessarily required, but is advantageous. Addition of an O/W emulsifier according to the invention or of two or more such emulsifiers in the coordinates L_2 or L_3 , on lowering the temperature, conveys the system to the coordinate L_4 , at which an O/W microemulsion according to the invention is present.
- 20

- A further advantageous embodiment of the process according to the invention accordingly consists, after choosing suitable raw materials, i.e. water phase and oil phase and optionally further substances, in bringing the individual components, with stirring, to a temperature at which phase inversion is possible for the given mixture and, by adding the O/W emulsifier used according to the invention or the O/W emulsifiers used according to the invention to the mixture, bringing about phase inversion, and afterwards allowing the mixture to cool to room temperature with continued stirring.
- 25
- 30

It is not beyond the ability of the person skilled in the art to determine, by simple experiments, the suitable temperature range within which a given mixture can pass

through phase inversion. This temperature range is usually to be chosen between 70 and 95°C, but in an individual case can also be above or below this.

5 In practice, it is possible and in some cases even advantageous for the temperature range which can be assigned to the phase inversion range also to be exceeded during the preparation of a microemulsion according to the invention since this range will then automatically be traversed upon cooling to room temperature.

10 The practice of the preparation of an emulsion according to the invention advantageously consists, after choosing suitable raw materials, i.e. water phase and oil phase, one or more emulsifiers of type A, the latter being present in concentrations at which phase inversion is possible for the given mixture, and optionally further substances, in heating the individual components with stirring to a temperature at which phase inversion is possible for the given mixture, and, by increasing or decreasing the pH of the mixture,
15 bringing about phase inversion, and afterwards allowing the mixture to cool to room temperature with continued stirring. One or more intermediate homogenization steps are advantageous, but are not absolutely necessary.

20 A further advantageous embodiment of the process according to the invention consists, after choosing suitable raw materials, i.e. water phase and oil phase, one or more emulsifiers of type A, the latter being present in concentrations at which phase inversion is possible for the given mixture, and optionally further substances, in bringing the individual components, with stirring, to a pH at which phase inversion is possible for the given mixture, and, by increasing the temperature of the mixture, bringing about phase
25 inversion, and afterwards allowing the mixture to cool to room temperature with continued stirring. One or more intermediate homogenization steps are advantageous, but are not absolutely necessary.

30 A third advantageous embodiment of the process according to the invention consists, after choosing suitable raw materials, i.e. water phase and oil phase, one or more emulsifiers of type A and optionally further substances, in bringing the individual components, with stirring, to a pH and a temperature at which phase inversion is possible for the given mixture, and, by adding the emulsifier A or the emulsifiers A to the mixture, bringing about phase inversion, and afterwards allowing the mixture to cool to room

temperature with continued stirring. One or more intermediate homogenization steps are advantageous, but are not absolutely necessary.

5 In practice, it is possible and in some cases even advantageous for the temperature range which can be assigned to the phase inversion range also to be exceeded during the preparation of an emulsion according to the invention since this range is then automatically traversed upon cooling to room temperature.

10 Cosmetic and dermatological preparations according to the invention comprise inorganic pigments, which are X-ray amorphous or non-X-ray amorphous, based on metal oxides and/or other metal compounds which are sparingly soluble or insoluble in water, in particular the oxides of titanium (TiO_2), zinc (ZnO), iron (e.g. Fe_2O_3), zirconium (ZrO_2), silicon (SiO_2), manganese (e.g. MnO), aluminum (Al_2O_3), cerium (e.g. Ce_2O_3), mixed oxides of the corresponding metals, and mixtures of such oxides. Particular preference is
15 given to pigments based on TiO_2 .

X-ray amorphous oxide pigments are metal oxides or semimetal oxides which reveal no or no recognizable crystal structure in X-ray diffraction experiments. Such pigments are often obtainable by flame reaction, for example by reacting a metal or semimetal halide
20 with hydrogen and air (or pure oxygen) in a flame.

In cosmetic dermatological or pharmaceutical formulations, X-ray-amorphous oxide pigments are used as thickeners and thixotropic agents, flow auxiliaries, for emulsion and dispersion stabilization and as carrier substance (for example for increasing the volume
25 of finely divided powders).

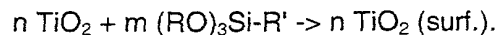
X-ray-amorphous oxide pigments which are known and are often used in cosmetic or dermatological technology are the silicon oxides of the Aerosil® grade (CAS No. 7631-86-9). Aerosils®, available from DEGUSSA, are characterized by low particle size (e.g.
30 between 5 and 40 nm), where the particles are to be regarded as spherical particles of very uniform dimension. Macroscopically, Aerosils® are recognizable as loose, white powders. Within the meaning of the present invention, X-ray-amorphous silicon dioxide pigments are particularly advantageous and, of these, precisely those of the Aerosil® grade are preferred.

Advantageous Aerosil® grades are, for example, Aerosil® OX50, Aerosil® 130, Aerosil® 150, Aerosil® 200, Aerosil® 300, Aerosil® 380, Aerosil® MOX 80, Aerosil® MOX 170, Aerosil® COK 84, Aerosil® R 202, Aerosil® R 805, Aerosil® R 812, Aerosil® R 972, Aerosil® R 974, Aerosil® R976.

According to the invention, cosmetic or dermatological light protection preparations advantageously comprise 0.1 to 20% by weight, advantageously 0.5 to 10% by weight, very particularly preferably 1 to 5% by weight, of X-ray-amorphous oxide pigments.

According to the invention, the non-X-ray-amorphous inorganic pigments are advantageously present in hydrophobic form, i.e. they have been surface-treated to repel water. This surface treatment can involve providing the pigments with a thin hydrophobic layer by methods known per se.

Such a method consists, for example, in producing the hydrophobic surface layer according to a reaction as in



n and m are stoichiometric parameters to be used as desired, and R and R' are the desired organic radicals. Hydrophobicized pigments prepared as in DE-A 33 14 742, for example, are advantageous.

Advantageous TiO_2 pigments are available, for example, under the trade names T 805 from Degussa.

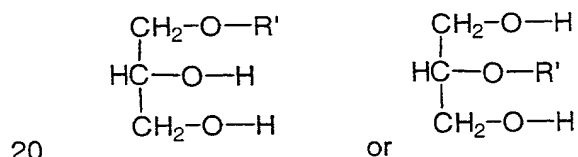
The total amount of inorganic pigments, in particular hydrophobic inorganic micropigments, in the finished cosmetic or dermatological preparations is advantageously chosen from the range 0.1 – 30% by weight, preferably 0.1 – 10.0% by weight, based on the total weight of the preparations.

The emulsifiers A are preferably chosen from the group of emulsifiers which are good proton donors or proton acceptors, it having to be ensured that either their lipophilicity is

dependent on the pH inasmuch as an increase or decrease in the pH results in an increase or decrease in lipophilicity, it being unimportant which of the two possibilities of change in the lipophilicity is effected by the increase or the decrease in the pH, or their lipophilicity is dependent on the temperature inasmuch as the lipophilicity increases with increasing temperature, and their hydrophilicity increases with decreasing temperature, or their lipophilicity is dependent on pH and temperature inasmuch as an increase or decrease in pH results in an increase or decrease in lipophilicity, it being unimportant which of the two possibilities of change in the lipophilicity is effected by the increase or the decrease in the pH, and inasmuch as the lipophilicity increases with increasing temperature and their hydrophilicity increases with decreasing temperature.

The emulsifiers of type A are advantageously chosen from the group of sorbitan esters and sucrose esters, in particular branched and unbranched alkyl esters and alkenyl esters having carbon chains of 4 - 24 carbon atoms, preferably sorbitan stearate, sorbitan oleate, glyceryl sorbitan stearate, sucrose monostearate, sucrose monolaurate, sucrose palmitate.

The emulsifiers of type A can advantageously be chosen from the group of monoglycerol monocarboxylic monoesters, in particular those characterized by the structures



where R' is a branched or unbranched acyl radical having 6 - 14 carbon atoms. R' is advantageously chosen from the group of unbranched acyl radicals.

The acids on which these esters are based are

25	hexanoic acid	(caproic acid)	(R' = -C ₅ H ₁₁),
	heptanoic acid	(enanthic acid)	(R' = -C ₆ H ₁₃),
	octanoic acid	(caprylic acid)	(R' = -C ₇ H ₁₅),
	nonanoic acid	(pelargonic acid)	(R' = -C ₈ H ₁₇),
	decanoic acid	(capric acid)	(R' = -C ₉ H ₁₉),
30	undecanoic acid		(R' = -C ₁₀ H ₂₁),
	10-undecenoic acid	(undecylenic acid)	(R' = -C ₁₀ H ₁₉),
	dodecanoic acid	(lauric acid)	(R' = -C ₁₁ H ₂₃),

tridecanoic acid (R' = -C₁₂H₂₅),
 tetradecanoic acid (myristic acid) (R' = -C₁₃H₂₇).

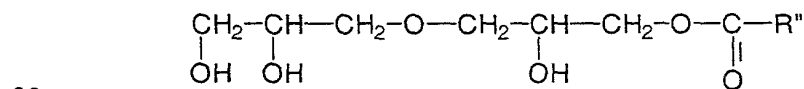
R' particularly advantageously represents the octanoyl radical (caprylic acid radical) or the decanoyl radical (capric acid radical), and is therefore represented by the formulae



The emulsifiers of type A can also be advantageously chosen from the group of di- and triglycerol monocarboxylic monoesters. According to the invention, the di- or triglycerol units of the diglycerol monocarboxylic monoesters or triglycerol monocarboxylic monoesters according to the invention are in the form of linear, unbranched molecules, i.e. "monoglycerol molecules" etherified via the respective OH groups in the 1- or 3-position.

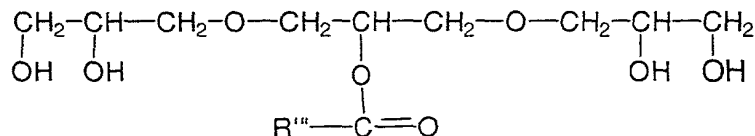
A low proportion of cyclic di- or triglycerol units, and glycerol molecules etherified via the OH groups in the 2-position can be tolerated. It is, however, advantageous to keep such impurities as low as possible.

The monocarboxylic monoesters according to the invention are preferably characterized by the following structure:



where R'' is a hydrocarbon radical, advantageously a branched or unbranched alkyl or alkenyl radical having 5 to 17 carbon atoms.

The monocarboxylic esters of triglycerol according to the invention are preferably characterized by the following structure:



where R''' is a hydrocarbon radical, advantageously a branched or unbranched alkyl or alkenyl radical having 5 to 17 carbon atoms.

The acids on which these esters are based are

	hexanoic acid	(caproic acid)	(R" and R"' = -C ₅ H ₁₁),
	heptanoic acid	(enanthic acid)	(R" and R"' = -C ₆ H ₁₃),
	octanoic acid	(caprylic acid)	(R" and R"' = -C ₇ H ₁₅),
	nonanoic acid	(pelargonic acid)	(R" and R"' = -C ₈ H ₁₇),
5	decanoic acid	(capric acid)	(R" and R"' = -C ₉ H ₁₉),
	undecanoic acid		(R" and R"' = -C ₁₀ H ₂₁),
	10-undecenoic acid	(undecylenic acid)	(R" and R"' = -C ₁₀ H ₁₉),
	dodecanoic acid	(lauric acid)	(R" and R"' = -C ₁₁ H ₂₃),
	tridecanoic acid		(R" and R"' = -C ₁₂ H ₂₅),
10	tetradecanoic acid	(myristic acid)	(R" and R"' = -C ₁₃ H ₂₇),
	pentadecanoic acid		(R" and R"' = -C ₁₄ H ₂₉),
	hexadecanoic acid	(palmitic acid)	(R" and R"' = -C ₁₅ H ₃₁),
	heptadecanoic acid	(margaric acid)	(R" and R"' = -C ₁₆ H ₃₃),
	octadecanoic acid	(stearic acid)	(R" and R"' = -C ₁₇ H ₃₅).

15

R" and R"' are particularly favorably chosen from the group of unbranched alkyl radicals having an uneven number of carbon atoms, in particular 9, 11 and 13 carbon atoms.

20

In general, the monocarboxylic monoesters of diglycerol are preferable to those of triglycerol.

According to the invention, very particular preference is given to

	diglycerol monocaprato	(DMC)	R" = 9
	triglycerol monolaurate	(TML)	R"' = 11
25	diglycerol monolaurate	(DML)	R" = 11
	triglycerol monomyristate	(TMM)	R"' = 13

30

A preferred monocarboxylic monoester of diglycerol according to the invention which has proven successful is diglycerol monocaprato (DMC).

In an advantageous embodiment of the present invention, an additional content of di- or triglycerol esterified in different positions is used, as is, where appropriate, a content of the various diesters of di- or triglycerol.

Also advantageous are triglyceryl diisostearate (nomenclature according to CTFA: polyglyceryl-3 diisostearate), isostearyldiglyceryl succinate, diglyceryl sesquiisostearate (nomenclature according to CTFA: polyglyceryl-2 sesquiisostearate), triglyceryl polyhydroxystearate (nomenclature according to CTFA: polyglyceryl-2 polyhydroxystearate).

5

Cetearyl isononanoate, dicocoylpentaerythrityldistearyl citrate, and also the methicone copolyols, cyclomethicone copolyols, alkylmethicone copolyols, in particular laurylmethicone copolyol, cetyldimethicone copolyol, have also proven advantageous according to the invention.

10

The emulsifier(s) of type A is/are particularly advantageously chosen from the group of branched or unbranched alkylmonocarboxylic acids, alkenylmonocarboxylic acids and alkylenedicarboxylic acids having 4 to 30 carbon atoms, in particular stearic acid, oleic acid, succinic acid, hexanoic acid (caproic acid), heptanoic acid (enanthic acid), octanoic acid (caprylic acid), nonanoic acid (pelargonic acid), decanoic acid (capric acid), undecanoic acid, undecenoic acid (undecylenic acid), dodecanoic acid (lauric acid), tridecanoic acid, tetradecanoic acid (myristic acid), pentadecanoic acid, hexadecanoic acid (palmitic acid), heptadecanoic acid (margaric acid), octadecanoic acid (stearic acid), isostearic acid, behenic acid. It is also advantageous to choose the emulsifiers A from the group of cosmetically or pharmaceutically acceptable salts of the abovementioned carboxylic acids, in particular the alkali metal, ammonium, monoalkylammonium, dialkylammonium, trialkylammonium and tetraalkylammonium salts.

15

20

The emulsifier(s) A is/are likewise particularly advantageously chosen from the group of mono-, oligo- and polyethoxylated compounds, in particular polyethoxylated mono- or polybasic alcohols or fatty acids, for example cetareth-20, PEG-20 glyceryl stearate, steareth-20, PEG-20 stearate, PEG-30 stearate, PEG-40 castor oil, PEG-1 glycerol sorbitan oleostearate, PEG-7 hydrogenated castor oil, PEG-40 sorbitan peroleate, PEG-45 dodecyl glycol copolymer.

25

30

The emulsions according to the invention advantageously comprise the emulsifier A or the emulsifiers A in concentrations of 0.01 – 20% by weight, preferably 0.05 – 10% by weight, particularly preferably 0.1 – 5% by weight, in each case based on the total weight of the composition.

According to the invention, it is possible to multiply the use amounts of UV filters which are themselves sparingly soluble or insoluble in oil components, in particular tris(2-ethylhexyl) 4,4',4''-(1,3,5-triazine-2,4,6-triyltriimino)trisbenzoate, but also 2-phenylbenzimidazole-5-sulfonic acid or salts thereof in cosmetic or dermatological preparations compared with the prior art.

The total amount of UV filter substances which are themselves sparingly soluble in oil components, in particular tris(2-ethylhexyl) 4,4',4''-(1,3,5-triazine-2,4,6-triyltriimino)trisbenzoate, but also 2-phenylbenzimidazole-5-sulfonic acid and salts thereof in the finished cosmetic or dermatological preparations is advantageously chosen from the range 0.1 – 10.0% by weight, preferably 0.5 – 6.0% by weight, based on the total weight of the preparations.

It is advantageous according to the invention to use additional oil-soluble UVA filters and/or UVB filters in the lipid phase and/or water-soluble UVA filters and/or UVB filters in the aqueous phase in the preparations according to the invention.

The light protection formulations according to the invention can advantageously comprise further substances which absorb UV radiation in the UVB region, the total amount of filter substances being, for example, 0.1% by weight to 30% by weight, preferably 0.5 to 10% by weight, in particular 1 to 6% by weight, based on the total weight of the preparations, in order to make available cosmetic preparations which protect the skin from the entire range of ultraviolet radiation.

25

The additional UVB filters can be oil-soluble or water-soluble. Examples of advantageous oil-soluble UVB filter substances are:

- 3-benzylidenecamphor derivatives, preferably 3-(4-methylbenzylidene)camphor, 3-benzylidenecamphor;
- 4-aminobenzoic acid derivatives, preferably 2-ethylhexyl 4-(dimethylamino)benzoate, amyl 4-(dimethylamino)benzoate;
- esters of cinnamic acid, preferably 2-ethylhexyl 4-methoxycinnamate, isopentyl 4-methoxycinnamate;

- derivatives of benzophenone, preferably 2-hydroxy-4-methoxybenzophenone, 2-hydroxy-4-methoxy-4'-methylbenzophenone, 2,2'-dihydroxy-4-methoxybenzophenone;
- esters of benzalmalonic acid, preferably di(2-ethylhexyl) 4-methoxybenzalmalonate;
- 5 - tris(2-ethylhexyl) 4,4',4''-(1,3,5-triazine-2,4,6-triyltriimino)trisbenzoate.

Examples of advantageous water-soluble UVB filter substances are:

- salts of 2-phenylbenzimidazole-5-sulfonic acid, such as its sodium, potassium or its triethanolammonium salt, and the sulfonic acid itself,
- 10 - sulfonic acid derivatives of benzophenones, preferably 2-hydroxy-4-methoxybenzophenone-5-sulfonic acid and salts thereof,
- sulfonic acid derivatives of 3-benzylidenecamphor, such as, for example, 4-(2-oxo-3-bornylidenemethyl)benzenesulfonic acid, 2-methyl-5-(2-oxo-3-bornylidenemethyl)sulfonic acid and salts thereof.

15

The list of said UVB filters which can be used in combination with the active ingredient combinations according to the invention is not of course intended to be limiting.

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It can also be advantageous to use additional UVA filters in the preparations according to the invention which have hitherto been customarily present in cosmetic preparations. These substances are preferably derivatives of dibenzoylmethane, in particular 1-(4'-tert-butylphenyl)-3-(4'-methoxyphenyl)propane-1,3-dione and 1-phenyl-3-(4'-isopropylphenyl)propane-1,3-dione. These combinations and preparations which comprise these combinations are also provided by the invention. The amounts which can be used are those

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used for the UVB combination.

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The cosmetic and/or dermatological light protection formulations according to the invention can have the customary composition and be used for cosmetic and/or dermatological light protection, and also for the treatment, care and cleansing of skin and/or hair and as a make-up product in decorative cosmetics.

For use, the cosmetic and dermatological preparations according to the invention are applied to the skin and/or hair in sufficient amount and in the manner conventional for cosmetics.

Particularly preferred cosmetic and dermatological preparations are those which are in the form of a sunscreen. Advantageously, these can additionally comprise at least one further UVA filter and/or at least one further UVB filter and/or at least one inorganic pigment, preferably an inorganic micropigment.

The cosmetic and dermatological preparations according to the invention can comprise cosmetic auxiliaries such as those conventionally used in such preparations, e.g. preservatives, bactericides, perfumes, antifoams, dyes, pigments which have a coloring effect, thickeners, moisturizers and/or humectants, fats, oils, waxes or other conventional constituents of a cosmetic or dermatological formulation, such as alcohols, polyols, polymers, foam stabilizers, electrolytes, organic solvents or silicone derivatives.

An additional content of antioxidants is generally preferred. According to the invention, favorable antioxidants which can be used are all antioxidants suitable or conventional for cosmetic and/or dermatological applications.

It is also advantageous to add antioxidants to the preparations according to the invention. The antioxidants are advantageously selected from the group consisting of amino acids (e.g. glycine, histidine, tyrosine, tryptophan) and derivatives thereof, imidazoles (e.g. urocanic acid) and derivatives thereof, peptides such as D,L-carnosine, D-carnosine, L-carnosine and derivatives thereof (e.g. anserine), carotenoids, carotenes (e.g. α -carotene, β -carotene, lycopene) and derivatives thereof, chlorogenic acid and derivatives thereof, lipoic acid and derivatives thereof (e.g. dihydrolipoic acid), aurothioglucose, propylthiouracil and other thiols (e.g. thioredoxin, glutathione, cysteine, cystine, cystamine and the glycosyl, N-acetyl, methyl, ethyl, propyl, amyl, butyl and lauryl, palmitoyl, oleyl, γ -linoleyl, cholesteryl and glyceryl esters thereof) and salts thereof, dilauryl thiodipropionate, distearyl thiodipropionate, thiodipropionic acid and derivatives thereof (esters, ethers, peptides, lipids, nucleotides, nucleosides and salts) and sulfoximine compounds (e.g. buthionine sulfoximines, homocysteine sulfoximine, buthionine sulfones, penta-, hexa-, heptathionine sulfoximine) in very low tolerated doses (e.g. pmol to μ mol/kg), and also (metal) chelating agents (e.g. α -hydroxy fatty acids, palmitic acid, phytic acid, lactoferrin), α -hydroxy acids (e.g. citric acid, lactic acid, malic acid), humic acid, bile acid, bile extracts, bilirubin, biliverdin, EDTA, EGTA and derivatives thereof, unsaturated fatty acids and derivatives thereof (e.g. γ -linolenic acid, linoleic acid, oleic acid), folic acid and derivatives thereof, ubiquinone and ubiquinol and

derivatives thereof, vitamin C and derivatives (e.g. ascorbyl palmitate, Mg ascorbyl phosphate, ascorbyl acetate), tocopherols and derivatives (e.g. vitamin E acetate), vitamin A and derivatives (vitamin A palmitate) and coniferyl benzoate of benzoin, rutinic acid and derivatives thereof, α -glycosylrutin, ferulic acid, furfurylidene-glucitol, carnosine, butylhydroxytoluene, butylhydroxyanisole, nordihydroguaiacic acid, nordihydroguaiaretic acid, trihydroxybutyrophenone, uric acid and derivatives thereof, mannose and derivatives thereof, zinc and derivatives thereof (e.g. ZnO, ZnSO₄), selenium and derivatives thereof (e.g. selenomethionine), stilbenes and derivatives thereof (e.g. stilbene oxide, trans-stilbene oxide) and the derivatives (salts, esters, ethers, sugars, nucleotides, nucleosides, peptides and lipids) of said active substances which are suitable according to the invention.

The amount of the abovementioned antioxidants (one or more compounds) in the preparations is preferably from 0.001 to 30% by weight, particularly preferably from 0.05 to 20% by weight, especially 1 - 10% by weight, based on the total weight of the preparation.

If vitamin E and/or derivatives thereof are used as the antioxidant or antioxidants, their respective concentrations are advantageously chosen from the range of 0.001 - 10% by weight, based on the total weight of the formulation.

If vitamin A or vitamin A derivatives or carotenes or derivatives thereof are used as the antioxidant or antioxidants, their respective concentrations are advantageously chosen from the range of 0.001 - 10% by weight, based on the total weight of the formulation.

The lipid phase can advantageously be chosen from the following group of substances:

- mineral oils, mineral waxes
- oils, such as triglycerides of capric or caprylic acid, but preferably castor oil;
- fats, waxes and other natural and synthetic fatty substances, preferably esters of fatty acids with alcohols of low carbon number, e.g. with isopropanol, propylene glycol or glycerol, or esters of fatty alcohols with alkanolic acids of low carbon number or with fatty acids;
- alkyl benzoates;
- silicone oils such as dimethylpolysiloxanes, diethylpolysiloxanes, diphenylpolysiloxanes and mixtures thereof.

For the purposes of the present invention, the oil phase of the emulsions, oleogels and hydrodispersions or lipodispersions is advantageously chosen from the group of esters of saturated and/or unsaturated, branched and/or unbranched alkanecarboxylic acids having a chain length of from 3 to 30 carbon atoms and saturated and/or unsaturated, branched and/or unbranched alcohols having a chain length of from 3 to 30 carbon atoms, from the group consisting of esters of aromatic carboxylic acids and saturated and/or unsaturated, branched and/or unbranched alcohols having a chain length of from 3 to 30 carbon atoms. Such ester oils can advantageously be selected from the group consisting of isopropyl myristate, isopropyl palmitate, isopropyl stearate, isopropyl oleate, n-butyl stearate, n-hexyl laurate, n-decyl oleate, isooctyl stearate, isononyl stearate, isononyl isononanoate, 2-ethylhexyl palmitate, 2-ethylhexyl laurate, 2-hexyldecyl stearate, 2-octyldodecyl palmitate, oleyl oleate, oleyl erucate, erucyl oleate, erucyl erucate and synthetic, semisynthetic and natural mixtures of such esters, e.g. jojoba oil.

The oil phase can also advantageously be chosen from the group of branched and unbranched hydrocarbons and hydrocarbon waxes, silicone oils, dialkyl ethers, from the group of saturated or unsaturated, branched or unbranched alcohols, and also fatty acid triglycerides, namely the triglycerol esters of saturated and/or unsaturated, branched and/or unbranched alkanecarboxylic acids of a chain length from 8 to 24, in particular 12 - 18, carbon atoms. The fatty acid triglycerides can advantageously be chosen, for example, from the group of synthetic, semisynthetic and natural oils, e.g. olive oil, sunflower oil, soybean oil, peanut oil, rapeseed oil, almond oil, palm oil, coconut oil, palm kernel oil and the like.

For the purposes of the present invention, any mixtures of such oil and wax components can also advantageously be used. When required, it can also be advantageous to use waxes, for example cetyl palmitate, as the sole lipid component of the oil phase.

The oil phase is advantageously chosen from the group consisting of 2-ethylhexyl isostearate, octyldodecanol, isotridecyl isononanoate, isoeicosane, 2-ethylhexyl cocoate, C₁₂₋₁₅-alkyl benzoate, caprylic/capric triglyceride and dicaprylyl ether.

Mixtures of C₁₂₋₁₅-alkyl benzoate and 2-ethylhexyl isostearate, mixtures of C₁₂₋₁₅-alkyl benzoate and isotridecyl isononanoate and mixtures of C₁₂₋₁₅-alkyl benzoate, 2-ethylhexyl isostearate and isotridecyl isononanoate are particularly advantageous.

Of the hydrocarbons, paraffin oil, squalane and squalene are advantageously to be used for the purposes of the present invention.

5 The oil phase can advantageously also contain cyclic or linear silicone oils or can consist entirely of such oils, although it is preferable to use an additional content of other oil phase components in addition to the silicone oil or silicone oils.

10 Cyclomethicone (octamethylcyclotetrasiloxane) is advantageously the silicone oil to be used according to the invention. However, other silicone oils can also advantageously be used for the purposes of the present invention, for example hexamethylcyclotrisiloxane, polydimethylsiloxane, poly(methylphenylsiloxane).

15 Mixtures of cyclomethicone and isotridecyl isononanoate and mixtures of cyclomethicone and 2-ethylhexyl isostearate are particularly advantageous.

The aqueous phase of the preparations according to the invention may advantageously comprise

20 - alcohols, diols or polyols of low carbon number, and also their ethers, preferably ethanol, isopropanol, propylene glycol, glycerol, ethylene glycol, ethylene glycol monoethyl or monobutyl ether, propylene glycol monomethyl, monoethyl or monobutyl ether, diethylene glycol monomethyl or monoethyl ether and analogous products, and also alcohols of low carbon number, e.g. ethanol, isopropanol, 1,2-propanediol, glycerol, and especially one or more thickeners which can
25 advantageously be chosen from the group consisting of silicon dioxide, aluminum silicates and polysaccharides and derivatives thereof, e.g. hyaluronic acid, xanthan gum, hydroxypropylmethylcellulose, and particularly advantageously from the group of polyacrylates, preferably a polyacrylate from the group consisting of the so-called carbopols, for example carbopol grades 980, 981, 1382, 2984, 5984,
30 in each case individually or in combination.

The text below briefly discusses some peculiarities and differences in the prerequisites of O/W emulsions and O/W microemulsions according to the invention.

35 Oils and fats differ inter alia in their polarity, which is difficult to define. It has already been proposed to adopt the interfacial tension with respect to water as a measure of the

polarity index of an oil or an oil phase. In this case, the lower the interfacial tension between this oil phase and water, the greater the polarity of the oil phase in question. According to the invention, the interfacial tension is to be regarded as one possible measure of the polarity of a given oil component.

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The interfacial tension is that force which acts on an imaginary line one meter in length located in the interface between two phases. The physical unit of this interfacial tension is conventionally calculated from the force/length relationship and is usually expressed in mN/m (millinewtons divided by meters). It has a positive sign if it endeavors to reduce the interface. In the converse case, it has a negative sign.

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According to the invention, the limit below which an oil phase is "polar" and above which an oil phase is "nonpolar" is regarded as 30 mN/m.

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According to the invention, the oil phase is advantageously chosen for O/W microemulsions from the group of polar oil components which have a polarity between 10 and 30 mN/m, where it must be ensured that at least one nonpolar oil component is present.

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Advantageous O/W microemulsions are obtained if the oil phase is chosen from the group of polar oil components, particularly preferably from the group of natural, synthetic or semisynthetic oil components, which have a polarity between 10 and 20 mN/m, where it must be ensured that at least one nonpolar oil component is present.

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It is also advantageous to use polar vegetable oils as polar oils of the O/W emulsions according to the invention. The vegetable oils can advantageously be chosen from the group of oils from the plant families Euphorbiaceae, Poaceae, Fabaceae, Brassicaceae, Pedalaceae, Asteraceae, Linaceae, Flacourtaceae, Violales, preferably chosen from the group consisting of natural castor oil, wheatgerm oil, grapeseed oil, kukui nut oil, safflower oil, thistle oil, oil of evening primrose and further oils which comprise at least 1.5% by weight of linoleic acid glycerides.

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The addition of electrolytes brings about a change in the solubility behavior of a hydrophilic emulsifier. The hydrophilic emulsifiers having the structures or properties described above pass through a partial phase inversion, leading to solubilization of water by the oil phase, which results in a stable microemulsion.

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The microemulsions according to the invention therefore advantageously comprise electrolytes, in particular one or more salts containing the following anions: chlorides, and also inorganic oxo element anions, and of these in particular sulfates, carbonates, phosphates, borates and aluminates. Electrolytes based on organic anions can also advantageously be used, for example lactates, acetates, benzoates, propionates, tartrates, citrates and many others. Comparable effects can also be achieved by ethylenediaminetetraacetic acid and salts thereof.

Cations of the salts which are preferably used are ammonium, alkylammonium, alkali metal, alkaline earth metal, magnesium, iron and zinc ions. It goes without saying that only physiologically acceptable electrolytes are to be used in cosmetics. On the other hand, specific medicinal applications of the microemulsions according to the invention may, at least in principle, require the use of electrolytes which should not be used without medical supervision.

Particular preference is given to potassium chloride, sodium chloride, magnesium sulfate, zinc sulfate and mixtures thereof. Also advantageous are salt mixtures as occur in the natural salt from the Dead Sea.

The concentration of the electrolyte or of the electrolytes should be about 0.01 – 10.0% by weight, particularly advantageously about 0.03 – 8.0% by weight, based on the total weight of the preparation.

The emulsifiers of type A are commonly regarded as O/W emulsifiers. A content of about 5 – 10% by weight of customary W/O emulsifiers advantageously promotes the formation of O/W/O emulsions, and a content of significantly more than 10% by weight of such emulsifiers leads to destabilization of the O/W/O emulsions.

If desired, for the preparation of O/W/O emulsions according to the invention, it is also advantageous to use hydrophilic and/or lipophilic gel formers. Although these do not generally contribute to the formation of multiple droplets, they promote the stability of multiple droplets once they have formed.

If, in a preparation process for O/W/O emulsions according to the invention, the pH is to be varied in order to bring an otherwise predetermined system into the phase inversion

range, then it is advantageous to initially use the lowest possible electrolyte concentration in the water phase at the start of the process, and if possible to initially dispense with such a concentration entirely. It is also advantageous to introduce emulsifier A into the oil phase, for example for stearic acid in the concentration range 0.5 – 5% by weight, in particular 2% by weight. The presence of emulsifier which is not covered by the definition of emulsifier A is advantageously in the concentration range from about 5 – 10% by weight, in particular about 7% by weight.

The pH should advantageously only be varied once the W/O emulsion has formed, for example by the addition of NaOH.

In this respect, it is within the general knowledge of the person skilled in the art and requires no inventive activity to determine the temperature and pH range in which phase inversion takes place for a given emulsifier or a given emulsifier system in a given water/oil phase system. As a general guideline for the PIT at customary emulsifier concentrations, a temperature range of about 40 - 90°C can be stated. In general, the PIT decreases as the emulsifier concentration increases.

If desired, during this process, the basic substances, auxiliaries, additives and/or active ingredients customary in cosmetics or medicinal galenics can also be added. It is clear to the person skilled in the art at which point in time such substances can be added to the process without the properties of the emulsion to be achieved being considerably impaired.

The examples below serve to outline the essence of the present invention in more detail without limiting the invention.

Example 1

	% by weight
Glycerol	5.00
Mineral oil	5.00
Glyceryl stearate, ceteareth-20, ceteareth-12, cetylstearyl alcohol, cetyl palmitate	3.50
C ₁₂₋₁₅ -alkyl benzoate	3.00
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	3.00
Ceteareth-20	1.50
Shea butter	1.00
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- 5 The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 2

	% by weight
Glycerol	5.00
Mineral oil	5.00
Glyceryl stearate, ceteareth-20, ceteareth-12, cetylstearyl alcohol, cetyl palmitate	3.50
C ₁₂₋₁₅ -alkyl benzoate	3.00
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	3.00
Ceteareth-20	1.50
Shea butter	1.00
Phenyltrimethicone	1.00
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- 5 The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 3

	% by weight
Glycerol	5.00
Glyceryl stearate, cetareth-20, cetareth-12, cetylstearyl alcohol, cetyl palmitate	3.70
Phenyltrimethicone	3.00
Mineral oil (Paraffinum liquidum)	3.00
Dicaprylyl ether	2.50
C ₁₂₋₁₅ -alkyl benzoate	2.50
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	2.00
Cetearyl isononanoate	1.50
Cetareth-20	1.30
Shea butter	0.75
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 4

	% by weight
Glycerol	5.00
Mineral oil	5.00
Glyceryl stearate, ceteareth-20, ceteareth-12, cetylstearyl alcohol, cetyl palmitate	3.50
C ₁₂₋₁₅ -alkyl benzoate	3.00
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	3.00
phenyltrimethicone	2.00
Ceteareth-20	1.50
Shea butter	1.00
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- 5 The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 5

	% by weight
Glycerol	5.00
Mineral oil	4.00
Glyceryl stearate, ceteareth-20, ceteareth-12, cetylstearyl alcohol, cetyl palmitate	3.70
Dicaprylyl ether	2.50
C ₁₂₋₁₅ -Alkylbenzoate	2.50
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	2.50
Phenyltrimethicone	1.50
Cetylstearyl isononanoate	1.50
Ceteareth-20	1.30
Shea butter	0.75
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 6

	% by weight
Glycerol	5.00
Mineral oil	5.00
Glyceryl stearate, ceteareth-20, ceteareth-12, cetylstearyl alcohol, cetyl palmitate	3.50
Shea butter	3.00
C ₁₂₋₁₅ -alkylbenzoate	3.00
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	3.00
Ceteareth-20	1.50
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- 5 The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Example 7

	% by weight
C ₁₂₋₁₅ -alkyl benzoate	4.50
Glyceryl stearate, cetareth-20, cetareth-12, cetylstearyl alcohol, cetyl palmitate	3.50
Mineral oil (Paraffinum liquidum)	3.50
Shea butter	3.00
Microcrystalline wax (Cera microcristallina), mineral oil (Paraffinum liquidum)	3.00
Glycerol	2.50
Cetareth-20	1.50
DMDM hydantoin	0.40
Perfume, dyes, preservatives	q.s.
Water	ad 100.00

- 5 The constituents of the oil phase are combined, as a result of which the mixture homogenizes, then combined with the water phase and brought to a temperature of 80-85°C (i.e. into the phase inversion temperature range of the system), then cooled to room temperature (i.e. brought out of the phase inversion temperature range of the system again).

Patent claims

1. An oil-in-water emulsion, in particular O/W microemulsion
- 5 (a) comprising at least one emulsifier (emulsifier A), chosen from the group of emulsifiers having the following properties
- their lipophilicity is either dependent on the pH inasmuch as an increase or decrease in pH results in an increase or decrease in lipophilicity, it being unimportant which of the two possibilities of change in the lipophilicity is effected by the increase or the decrease in the pH, and/or
- 10 - their lipophilicity is dependent on the temperature inasmuch as the lipophilicity increases with increasing temperature and their hydrophilicity increases with decreasing temperature,
- (b) also optionally further substances which are soluble or dispersible in the oil phase or the water phase, including, preferably, those chosen from the group of
- 15 emulsifiers not covered by the definition of emulsifier A, in particular those which act primarily as W/O emulsifiers,
- (c) an effective amount of shea butter.
2. The O/W macroemulsion or O/W microemulsion as claimed in claim 1, wherein the
- 20 emulsifier A or the emulsifiers A is or are present in concentrations of 0.01-20% by weight, preferably 0.05-10% by weight, particularly preferably 0.1-5% by weight, in each case based on the total weight of the composition.
3. The O/W macroemulsion or O/W microemulsion as claimed in claim 1, which
- 25 comprises 0.1 to 20% by weight, advantageously 0.5 to 10% by weight, very particularly preferably 1 to 5% by weight, of shea butter.

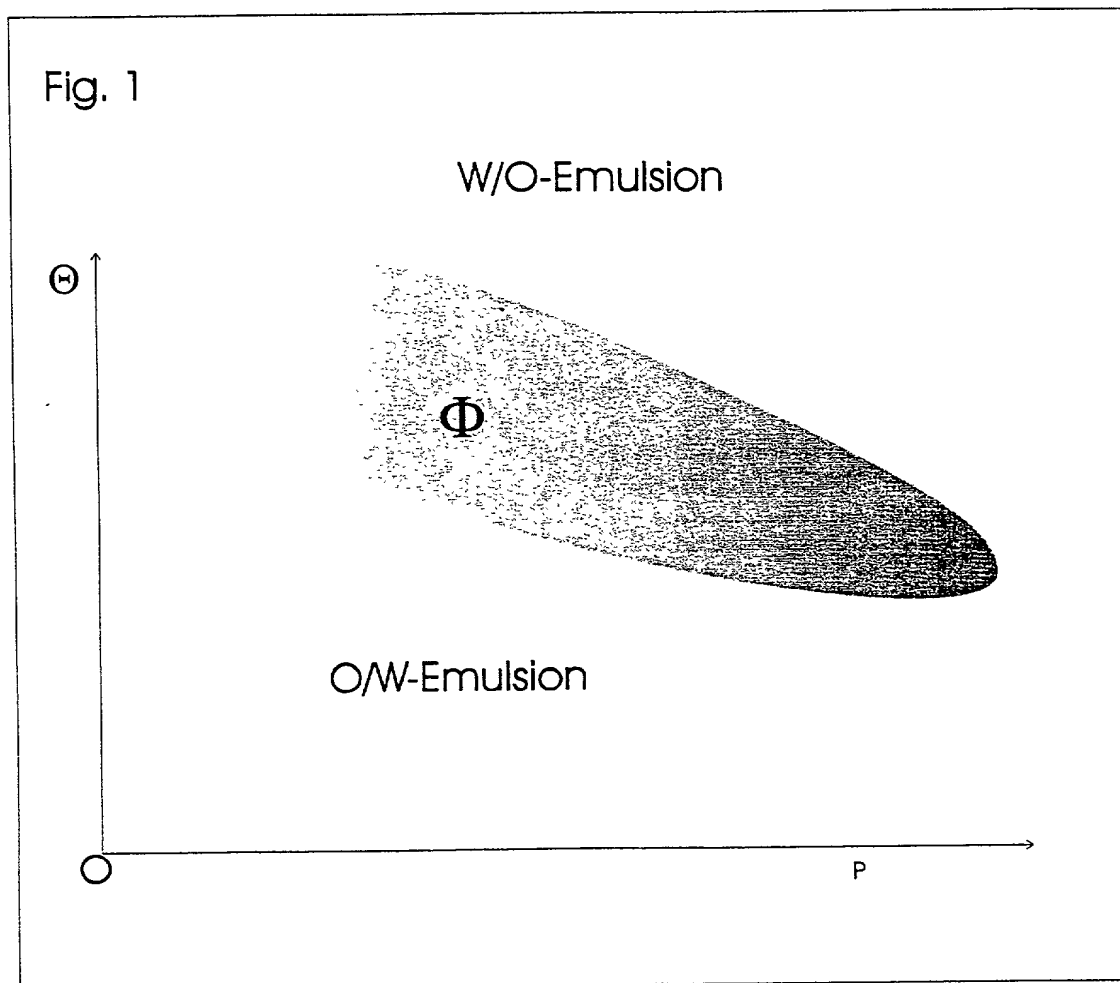
Abstract

Sprayable oil-in-water emulsions, in particular O/W microemulsions, comprising inorganic pigments [lacuna] emulsifiers whose lipophilicity is dependent either on the pH or on the temperature, and one or more film formers chosen from the group of silicone emulsifiers.

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Fig. 1



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Fig. 2

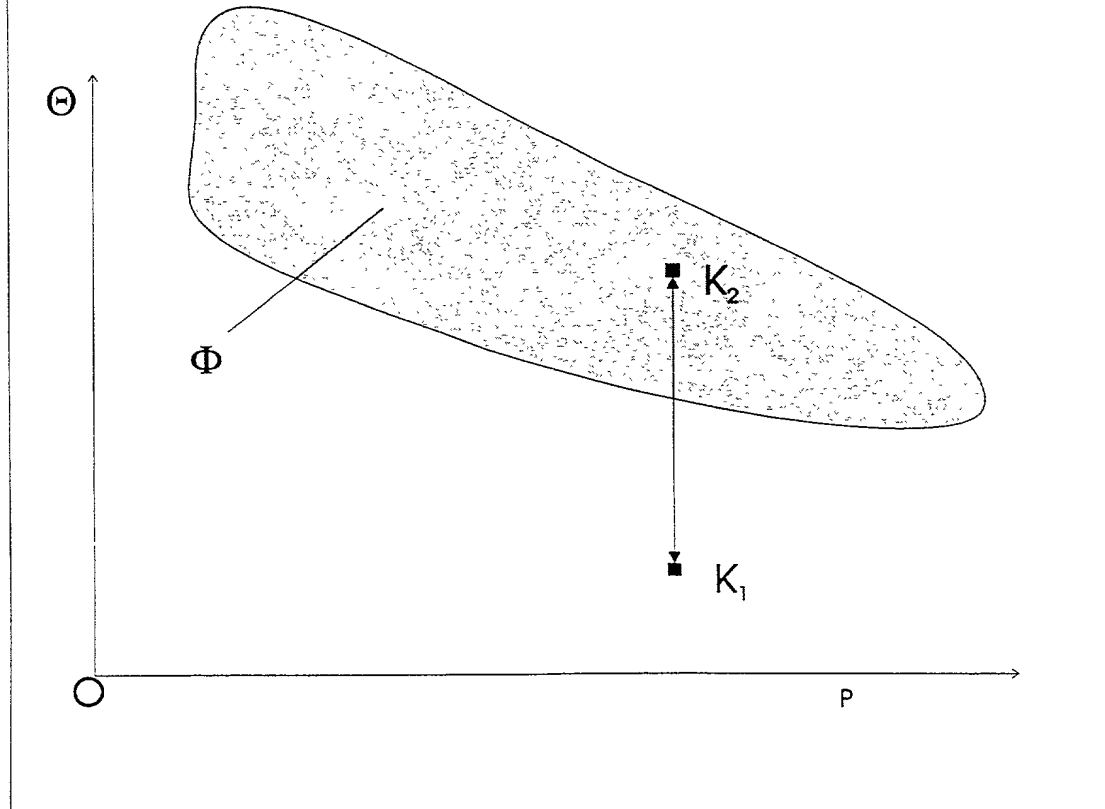
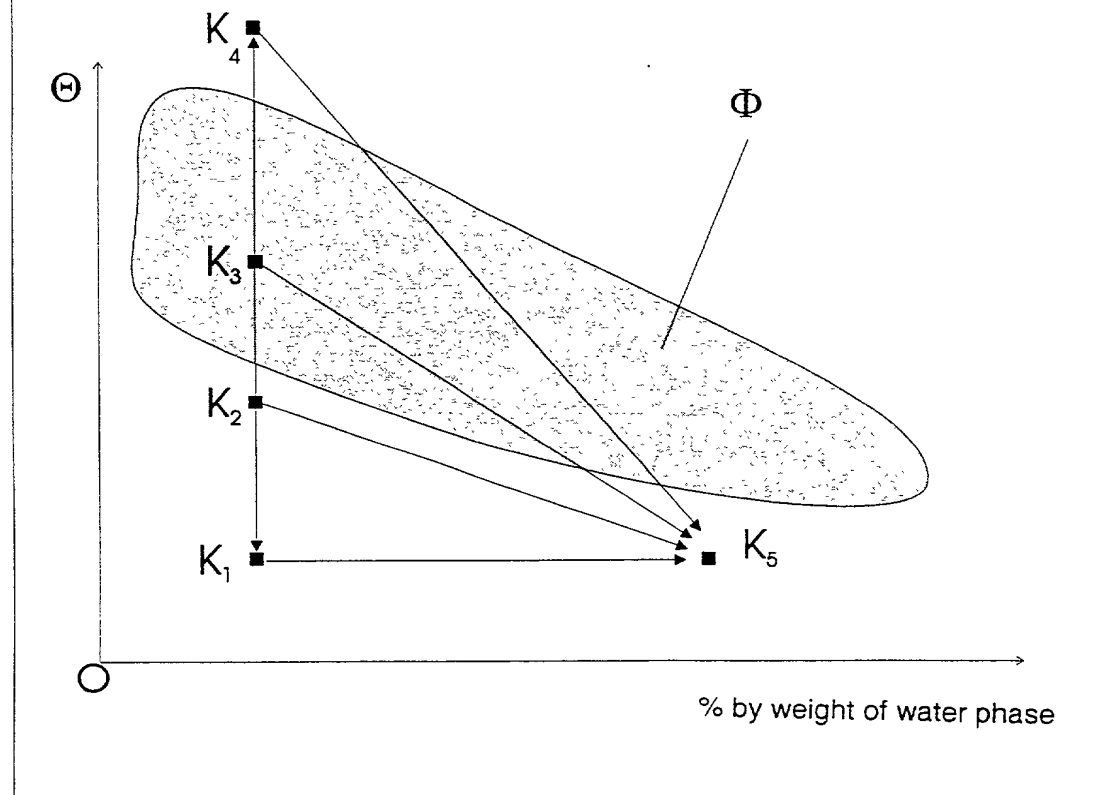
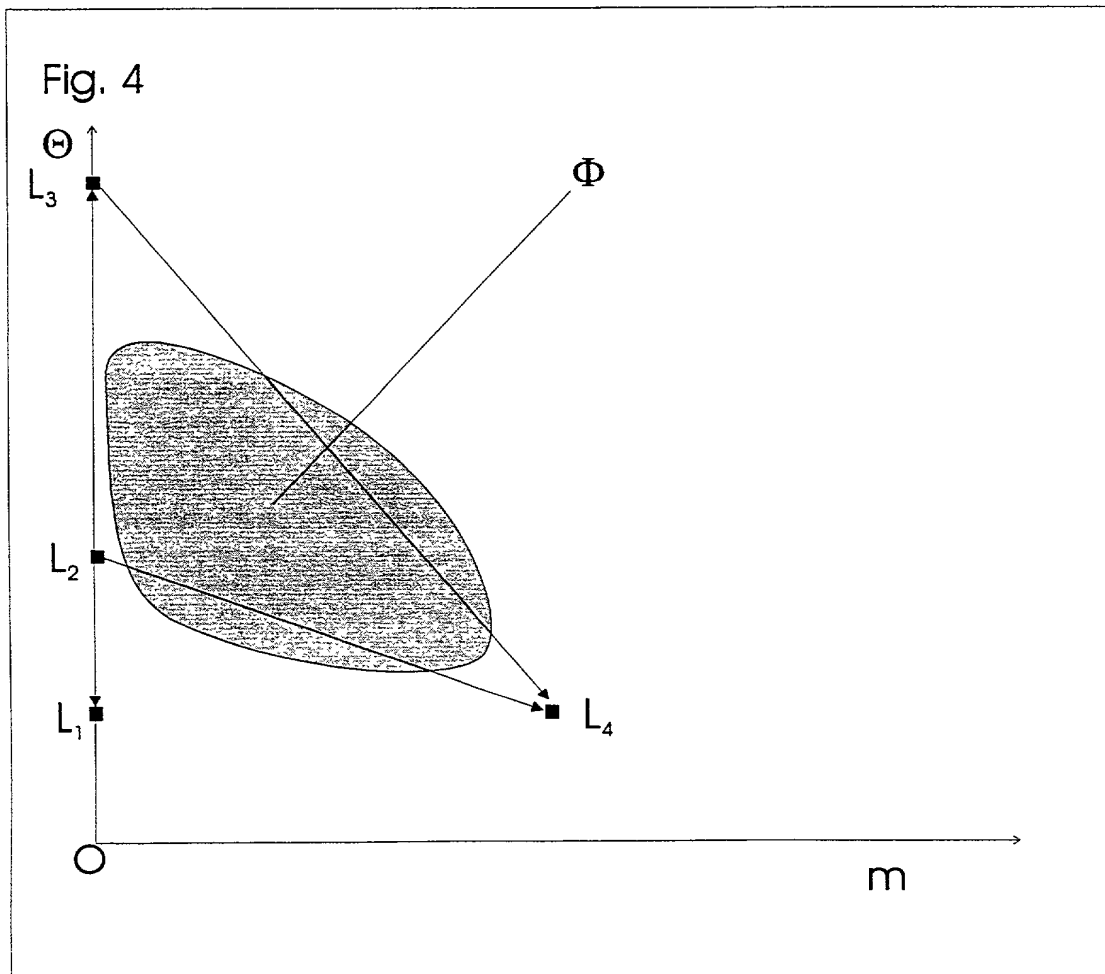


Fig. 3



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COMBINATION DECLARATION & POWER OF ATTORNEY

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name,

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled **„Cosmetic and dermatological light protection formulations in the form of O/W macroemulsions or O/W microemulsions containing shea butter“** the specification of which is attached hereto.

-OR-

was filed on _____ as

Application Serial No. _____ and was amended _____

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations §1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code, §119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

Prior Foreign Application(s)	Priority Claimed
<u>199 57 761.7</u> (Number)	<u>Germany</u> (Country)
<u>01/12/1999</u> (Day/Month/Yr. Filed)	<input checked="" type="checkbox"/> yes <input type="checkbox"/> no
<u> </u> (Number)	<u> </u> (Country)
<u> </u> (Day/Month/Yr. Filed)	<input checked="" type="checkbox"/> yes <input type="checkbox"/> no

I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

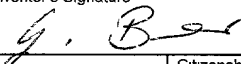
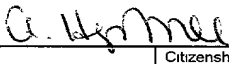

<u> </u> (Application Serial No.)	<u> </u> (Filing Date)	<u> </u> (Status)
		(patented, pending, abandoned)

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punished by fine or imprisonment, or both under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

POWER OF ATTORNEY: As a named Inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith:

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